STUDY OF POSITIVE ELECTRODE AND SEPARATOR FACTORS AFFECTING PERFORMANCE AND LIFE OF LITHIUM/IRON SULFIDE CELLS

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Parametric design studies are being conducted to improve the utilization of the FeS₂ electrode in Li-Al/LiCl-KCl/FeS₂ cells. The parameters under investigation include: (1) electrode thickness; (2) electrode porosity; (3) current collection; (4) electrolyte composition. The utilization of active material in the FeS₂ electrode is determined by observations of the voltage vs. time characteristics of small (<10 cm²) test electrodes under constant current discharge. The electrode potentials are measured relative to a suitable reference electrode. The effects of current density and temperature are also being evaluated.

The results of the studies to date indicate that (1) a porosity of 30% for the uncharged electrode is near the optimum, (2) an electrode thickness of \sim 3 mm is near the optimum, and (3) the use of 3 - 4 wt.% Vulcan XC72R graphite powder is near the optimum as a current-collector material. Variations in temperature (400 and 450 °C) and current density (2- and 4-h discharge rates) had only a slight effect on the electrode behavior. Of the three parameters investigated to date, improved current collection had the greatest effect in increasing the utilization of the FeS₂ electrode. Studies to evaluate other current collectors and the effect of electrolyte composition are under consideration.

DEVELOPMENT OF *IN SITU* BORONIZATION PROTECTION OF POSITIVE ELECTRODE CURRENT COLLECTORS

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The concept of *in situ* boronization protection for the positive electrode current collector in lithium/metal sulfide batteries recognizes that corrosionresistant coatings may be susceptible to corrosion of the substrate *via* defects or pinholes. Boride coatings on low-carbon ferrous materials, although showing excellent initial corrosion resistance for this application, may become more susceptible to corrosion *via* slow diffusion of the boride from the surface into the substrate. Such diffusional losses and coating defects can be compensated for by introducing iron borides (Fe₂B and FeB) into the positive active material where they will be available for diffusion to the initially boronized material.

The objective of this program was to give a preliminary demonstration of the basic concept. High corrosion stabilities were found for low-carbon ferrous materials boronized by the IGT pack boronization technique. Boride migration from iron boride-containing positive active material has been demonstrated, which, under long-term battery cycling, is expected to enhance the practical lifetime of the initial coating. Reproducible pack boronization techniques have been developed which allow for the rapid boronization of lithium-metal sulfide components and are readily amenable to commercial scale-up. No cell assembly difficulties were identified as a result of this technique.

Recent publications

1 M. R. St. John and A. F. Sammells, Development of *in situ* boronization protection of positive electrode current collectors, Final Report, June 1979.

INSULATION DEVELOPMENT FOR HIGH-TEMPERATURE BATTERY FOR ELECTRIC VEHICLE APPLICATION

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The objective of this contract is to develop and demonstrate a highperformance vacuum insulation which is capable of withstanding 0.1 MPa (15 psi) plus battery loading with low compression, operating in the 350 -450 °C range. The developed insulation would allow construction of rectangular, lightweight and low-cost, vacuum-insulated enclosures for electric vehicles using Na/S or Li/MS batteries. The goals of the program are to develop a vacuum insulation with the following properties: thermal conductivity 0.0024 - 0.0035 W/m K ($140 \times 10^{-5} - 200 \times 10^{-5}$ BTU/h ft °F); density 288 kg/m³ (18 lb/ft³); compression 10% from 0 to 0.103 MPa (15 psi) load.

A new milestone in high-temperature, load-bearing, preformed insulation has been achieved through Linde's efforts on this program. Prior to this program, the best load-bearing commercial vacuum insulation system had a thermal conductivity six times higher than the goal.